

Georgia Institute of Technology

**The George W. Woodruff School of Mechanical Engineering
Nuclear & Radiological Engineering/Medical Physics Program**

Ph.D. Qualifier Exam

Fall Semester 2010

_____ Your ID Code

Radiation Physics (Day 1)

Instructions

1. Use a separate page for each answer sheet (no front to back answers).
2. The question number should be shown on each answer sheet.
3. ANSWER 4 OF 6 QUESTIONS ONLY.
4. Staple your question sheet to your answer sheets and turn in.

NRE/MP Radiation Physics

Answer any 4 of the following 6 questions.

- Q1. At $t=0$, there are exactly N_0 hypothetical radioactive nuclei. At $t = T$ (exactly), N_T nuclei are left. The decay constant is calculated in terms of N_0 , N_T and T .
- Estimate the standard deviation of the calculated decay constant.
 - Estimate the optimal T that minimizes the standard deviation of the calculated decay constant.
- Q2. If the Q value for the reaction ${}^3\text{He}(n,p){}^3\text{H}$ is 0.764 MeV, what is the threshold energy for the reaction ${}^3\text{H}(p,n){}^3\text{He}$?
- Different from the reaction ${}^3\text{He}(n,p){}^3\text{H}$, the cross section of the reaction ${}^3\text{H}(p,n){}^3\text{He}$ is largely affected by the Coulomb force. Estimate the Coulomb barrier for the reaction ${}^3\text{H}(p,n){}^3\text{He}$, assuming the equivalent hard-sphere radius of the proton is 1.05 fm.
- Q3. Consider the 3 component decay chain $X_1 \xrightarrow{\lambda_1} X_2 \xrightarrow{\lambda_2} X_3$.
- Derive an expression for the amount and the activities of X_2 and X_3 at any time t as a function of decay constants. Assume at time $t=0$, only N_1^0 amount of X_1 is present.
 - When will these radionuclides be in secular equilibrium? You may state the condition for secular equilibrium in terms of decay constants.
- Q4. The isotope ${}^{18}\text{F}$ is a radionuclide used in medical diagnoses of tumors and, although usually produced by the ${}^{18}\text{O}(p,n){}^{18}\text{F}$ reaction, it can also be produced by irradiating lithium carbonate (Li_2CO_3) with neutrons. The neutrons interact with ${}^6\text{Li}$ to produce tritons (nuclei of ${}^3\text{H}$) which in turn interact with the oxygen to produce ${}^{18}\text{F}$.
- What are the two nuclear reactions?
 - Calculate the Q-value for each reaction
 - Calculate the threshold energy for each reaction
 - Can thermal neutrons be used to create ${}^{18}\text{F}$?
- Q5. Calculate the maximum energy that a 10-MeV electron can possibly transfer to an atomic electron? In your calculation, please show the energy conservation and momentum conservation equations for the interacting electrons.

NRE/MP Radiation Physics – Cont'd.

- Q6.** In an Inverse-Compton Scattering experiment (shown in the figure below), a 10-MeV electron beam and a 3-eV laser beam are brought head-to-head toward each other. What will be the energy of the photons scattered into the 60° direction?

